# C-Nucleosides. 11.t Synthesis of Quinoxaline C-Nucleosides through Condensation of 1,2-Diaminobenzenes with 6-Hydroxy-6-(2,3,5-tri-O-benzoyl- $\beta$ -D-ribofuranosyl)-2,6-dihydropyran-3-one 

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#### Abstract

The synthesis of 6 - and 7 -substituted-2-( $\beta$ - D -ribofuranosyl) quinoxaline and 7 - and 8 -substituted-1( $\beta$ - D-ribofuranosyl) pyrrolo[1,2-a]quinoxaline from 6 -hydroxy-6-(2,3,5-tri-O-benzoyl- $\beta$-D-ribofuran-osyl)-2,6-dihydropyran-3-one (1) is described. Treatment of (1) with 1,2-diamino-4-chlorobenzene (2a) afford three compounds, the 6 - and 7 -chloroquinoxalines (3a) and (3b) and the 7 -chloropyrrolo 1,2 -a]quinoxaline (4a) in 23, 43, and $9 \%$ yield, respectively. The position of the substituent in products (3a) and (3b) was determined by comparison of these ${ }^{1} \mathrm{H}$ n.m.r. spectra with those of the corresponding $N$-oxides (5a), (6a), and (5b), (6b), prepared by oxidation of compounds (3a) and (3b) with $m$-chloroperbenzoic acid. The position of the substituent in (4a) was confirmed by ${ }^{1} \mathrm{H}-{ }^{13} \mathrm{C}$ long-range COSY experiment with corresponding deblocked pyrrolo[1,2a]quinoxaline ( $\mathbf{4 c}$ ). Treatment of compound (1) with 1,2-diamino-4-nitrobenzene ( $\mathbf{2 b}$ ) afforded two compounds, the 6 -nitroquinoxaline (3c) and the 8 -nitropyrrolo[1,2-a]quinoxaline (4b). Deprotection of compounds (3a-c), (4a, b), (5a, b), and (6a, b) with methanolic sodium hydroxide afforded ( $\mathbf{3 d}-\mathbf{f}$ ), ( $4 \mathbf{c}, \mathbf{d}$ ), ( $5 \mathbf{c}, \mathbf{d}$ ), and ( $6 \mathbf{c}, \mathbf{d}$ ), respectively.


In a recent report from our laboratory, we described the preparation of a functionalized $C$-glycoside, 6 -hydroxy-6-(2,3,5-tri-$O$-benzoyl- $\beta$-D-ribofuranosyl)-2,6-dihydropyran-3-one (1), and its utilization in the synthesis of quinoxaline and pyrrole $C$ nucleosides. ${ }^{1}$ We have investigated a convenient and general synthesis of quinoxaline derivatives through condensation of 1,2-diaminobenzenes with compound (1). It was of interest to examine whether 6 -substituted quinoxaline $C$-nucleosides or the corresponding 7 -isomers would be obtained on treatment of 1,2 -diamino-4-substituted benzenes with compound (1). The key synthetic intermediate pyranulose (1) can be obtained readily from 2 -( $2,3,5$-tri- $O$-benzoyl- $\beta$-D-ribofuranosyl)furan by our previously published procedure. ${ }^{1}$
Treatment of 1,2-diamino-4-chlorobenzene (2a) with pyranulose (1) in chloroform under reflux gave a mixture of 6-chloro-2( $2,3,5$-tri- $O$-benzoyl- $\beta$-d-ribofuranosyl)quinoxaline (3a), 7-chloro-2-( $2,3,5$-tri- $O$-benzoyl- $\beta$-D-ribofuranosyl)quinoxaline (3b), and 7 -chloro-1-(2,3,5-tri- $O$-benzoyl- $\beta$-d-ribofuranosyl)pyrrolo $[1,2-a]$ quinoxaline (4a) in 23,43 , and $9 \%$ yield, respectively (Scheme 1). The position of the substituent in compounds ( $\mathbf{3 a}$ ) and ( $\mathbf{3 b}$ ) was determined by comparison of their ${ }^{1} \mathrm{H}$ n.m.r. spectra with those of the corresponding $N$-oxides (5a), (6a), and ( $\mathbf{5 b}$ ), ( $6 \mathbf{b}$ ), prepared by oxidation of (3a) and (3b) with $m$-chloroperbenzoic acid (MCPBA). The ratios of 1-oxide (5a)/4-oxide (6a) and 1 -oxide ( $\mathbf{5 b}$ )/4-oxide ( $\mathbf{6 b}$ ) were $\sim 1: 3$, respectively. The position of the $\mathrm{N}-\mathrm{O}$ group in these N -oxides was also determined by analysis of the ${ }^{1} \mathrm{H}$ n.m.r. spectra of the corresponding $N$-oxides. In ${ }^{1} \mathrm{H}$ n.m.r. spectra, the signals for the proton adjacent to the $N$-oxide underwent an upfield shift relative to the parent quinoxaline, whereas the signals due to the proton at the peri-position to the $N$-oxide were displaced downfield. ${ }^{2}$ The signal of $3-\mathrm{H}$ of 4 -oxide ( $6 \mathbf{a}$ ) at $\delta 8.61$ occurs at higher field than that of (3a) at $\delta 9.06$, and 3 -H of 1 -oxide (5a) shifted to $\delta 9.04$. These data indicate that (5a) and (6a) are the 1oxide and 4 -oxide, respectively. The spectra of compounds (5a) and (6a) contained doublets at $\delta 8.47(J 9.4 \mathrm{~Hz})$ and $8.51(J 2.0$

[^0]Hz ), which could be assigned to $8-\mathrm{H}$ and $5-\mathrm{H}$ at the position peri to the $N$-oxide. These coupling constants indicated that the chloro group was located at the 6 -position. As summarized in Table 1, (5b) and ( $\mathbf{6 b}$ ) are the 1 -oxide and 4 -oxide, respectively. The chloro group was located at the 7-position as follows. The position of the substituent in the pyrrolo [1,2-a]quinoxaline (4a) was confirmed by a ${ }^{1} \mathrm{H}-{ }^{13} \mathrm{C}$ long-range COSY experiment with the corresponding deblocked pyrrolo [1,2-a]quinoxaline (4c). In ${ }^{1} \mathrm{H}-{ }^{13} \mathrm{C}$ long-range experiments of $(\mathbf{4 c})$, a correlation was observed between 9-H at $\delta 8.46(J 9.1 \mathrm{~Hz})$ and C-5a and C-7 at $\delta_{\mathrm{C}} 137.93$ and 129.06. Other long-range correlations are shown by arrows in the Figure. This data indicated that the chloro group was located at the 7 position. Coupling constants for the ${ }^{1} \mathrm{H}$ n.m.r. spectra are given in Table 2.
Next, the reaction between 1,2-diamino-4-nitrobenzene ( $\mathbf{2 b}$ ) and the pyranulose (1) afforded 6-nitro-2-(2,3,5-tri- $O$-benzoyl-$\beta$-d-ribofuranosyl)quinoxaline (3c) in $40 \%$ yield without formation of the 7-nitroquinoxaline isomer, and small amounts of 8 -nitro-1-(2,3,5-tri- $O$-benzoyl- $\beta$-D-ribofuranosyl)pyrrolo-[1,2-a]quinoxaline (4b). In order to determine the position of nitro group, we attempted to prepare the corresponding $N$ oxide compound. However, attempted $N$-oxidation of (3c) with MCPBA resulted in the recovery of unchanged starting material. The ${ }^{1} \mathrm{H}^{13} \mathrm{C}$ long-range COSY spectrum of compound (3c) exhibited a correlation between $8-\mathrm{H}$ at $\delta 8.09$ ( $J$ 9.2 Hz ) and $\mathrm{C}-4 \mathrm{a}$ at $\delta_{\mathrm{C}} 141.25$. Other long-range correlations are shown by arrows in the Figure. Hence, the nitro group was located at C-6. The position of the nitro group in (4b) was confirmed by a ${ }^{1} \mathrm{H}_{-}{ }^{13} \mathrm{C}$ COSY experiment with the corresponding deblocked pyrrolo[1,2- $a$ ]quinoxaline (4d). The spectrum of compound ( $\mathbf{4 d}$ ) contained a doublet at low field $\delta$ $9.60(J 2.3 \mathrm{~Hz}, 9-\mathrm{H}){ }^{3}$ This coupling constant indicated that the nitro group was located at $\mathrm{C}-8$. This positionally selective synthesis of compounds ( $\mathbf{3 c}$ ) and (4b) most probably proceeds via preferential reaction of C-5 and C-3 in the pyranulose with the more basic amino group in (2b) (Scheme 2). The preponderant product ( $\mathbf{3 b}$ ) would result from preferential reaction at $\mathrm{C}-5$ in the pyranulose by the more basic amino group in (2a).


$+$


|  | $\mathrm{X}^{1}$ | $\mathrm{X}^{2}$ | $\mathrm{R}^{\prime}$ | R |
| ---: | :--- | :--- | :--- | :--- |
| (3)a; | Cl | H | Bz | Bz |
| $\mathbf{b} ;$ | H | Cl | Bz | Bz |
| $\mathbf{c} ;$ | $\mathrm{NO}_{2}$ | H | Bz | Bz |
| $\mathbf{d} ;$ | Cl | H | H | H |
| $\mathbf{e} ;$ | H | Cl | H | H |
| $\mathbf{f} ;$ | $\mathrm{NO}_{2}$ | H | H | H |
| $\mathbf{g} ;$ | $\mathrm{Cl}^{2}$ | H | H | $\mathrm{R}=\mathrm{CMe}_{2}$ |
| $\mathbf{h} ;$ | H | Cl | H | $\mathrm{R}=\mathrm{CMe}_{2}$ |
| $\mathbf{i} ;$ | $\mathrm{NO}_{2}$ | H | H | $\mathrm{R}=\mathrm{CMe}_{2}$ |


|  | $\mathrm{X}^{1}$ | $\mathrm{X}^{2}$ | $\mathrm{R}^{\prime}$ | R |
| ---: | :--- | :--- | :--- | :--- |
| (4)a; | Cl | H | Bz | Bz |
| b; | H | $\mathrm{NO}_{2}$ | Bz | Bz |
| c; | Cl | H | H | H |
| d; | H | $\mathrm{NO}_{2}$ | H | H |
| e; | Cl | H | H | $\mathrm{R}=\mathrm{CMe}_{2}$ |
| f; | H | $\mathrm{NO}_{2}$ | H | $\mathrm{R}=\mathrm{CMe}_{2}$ |

Scheme 1.



|  | $X^{1}$ | $X^{2}$ | $R$ |
| :---: | :--- | :--- | :--- |
| (5) $\mathbf{a} ;$ | Cl | H | Bz |
| $\mathbf{b} ;$ | H | Cl | Bz |
| $\mathbf{c} ;$ | Cl | H | H |
| $\mathbf{d} ;$ | H | Cl | H |


|  | $\mathrm{X}^{1}$ | $\mathrm{X}^{2}$ | R |
| ---: | :--- | :--- | :--- |
| (6) $\mathbf{a} ;$ | Cl | H | Bz |
| $\mathbf{b} ;$ | H | Cl | Bz |
| c; | Cl | H | H |
| d; | H | Cl | H |

Deprotection of compounds ( $\mathbf{3 a - c}$ ), ( $\mathbf{4 a}, \mathbf{b}$ ), $(5 a, b)$, and ( $\mathbf{6 a}$, b) with methanolic sodium hydroxide afforded products (3df), ( $4 \mathbf{c}, \mathrm{~d}$ ), ( $5 \mathrm{c}, \mathrm{d}$ ), and ( $\mathbf{6 c}, \mathrm{d}$ ), respectively. The assignments of anomeric configurations of compounds ( $\mathbf{3 d}-\mathbf{f}$ ) and ( $\mathbf{4 c}, \mathbf{d}$ ) were made on the basis of the difference in the chemical shifts of the two methyl signals of the corresponding 2,3-O-isopropylidene derivatives ( $\mathbf{3 g - i}$ ) and ( $\mathbf{4 e}, \mathbf{f}$ ). The ${ }^{1} \mathrm{H}$ n.m.r. chemical-shift

(3c)

$R=$


(4c)


Figure The ${ }^{1} \mathrm{H}^{-13} \mathrm{C}$ long-range COSY experiments with compounds (3c) and (4c)
differential value ( $\Delta \delta$ ) of the methyl groups in the isopropylidene derivatives is indicative of $\beta$ stereochemistry in accordance with the Imbach's rule ( $<0.15$ and $>0.15$ p.p.m. for the $\alpha$ and $\beta$ anomers) ${ }^{4}$ (see Experimental Section). Deoxygenation of $N$-oxides ( $\mathbf{5 c}$ ), ( $\mathbf{6 c}$ ), ( 5 d ), and ( $\mathbf{6 d}$ ) with triphenylphosphine in methanol gave (3d) and (3e), respectively. This showed that the $\beta$-ribofuranoside configuration had been preserved during the reaction sequence.

## Experimental

M.p.s were determined on a Yanagimoto apparatus and are uncorrected. Mass spectra were taken on a Hitachi M-80 instrument by direct insertion at 70 eV , fast-atom bombardment (f.a.b.) mass spectra were run on a JMS-HX 110 using nitrobenzyl alcohol. ${ }^{1} \mathrm{H}$ N.m.r. spectra were measured with a JNM-GX-270 and a GX-400 (JEOL) spectrometer, with tetramethylsilane as internal standard. ${ }^{13} \mathrm{C}$ N.m.r. spectra were recorded on a JEOL JNM-FX-100 Fourier transform spectrometer operating at 25.00 MHz , with tetramethylsilane as internal standard. Elemental analysis were determined by the analytical centre of this faculty. Analytical t.l.c. was performed on glass plates coated with a $0.5-\mathrm{mm}$ layer of silica gel $\mathrm{GF}_{254}$ (Merck). The compounds were detected by u.v. light ( 254 nm ). Column chromatography was performed on silica gel C-200 (74-149 $\mu \mathrm{m}$, Wakogel).

6-Chloro- and 7-Chloro-2-(2,3,5-tri-O-benzoyl- $\beta$-D-ribofuranosyl)quinoxaline (3a) and (3b) and 7-Chloro-1-(2,3,5-tri-O-benzoyl- $\beta$-D-ribofuranosyl)pyrrolo[1,2-a]quinoxaline (4a).-A solution of the pyranulose (1) ( $825 \mathrm{mg}, 1.48 \mathrm{mmol}$ ) and $1,2-$ diamino-4-chlorobenzene ( $\mathbf{2 a}$ ) ( $253 \mathrm{mg}, 1.77 \mathrm{mmol}$ ) in chloroform ( 10 ml ) was heated under reflux for 5 h , and then the solvent was evaporated off under reduced pressure. T.l.c. (chloroform-methanol, 9:1) showed that the light yellow syrup contained three major components ( $R_{\mathrm{F}} 0.15,0.17$, and 0.18 ). The mixture was separated by preparative t.l.c. (p.l.c.) with chloroform as developer ( $\times 6$ ).

Compound (3a) ( $212 \mathrm{mg}, 23 \%$ ); $R_{\mathrm{F}} 0.18$; syrup (Found: C, 66.8; $\mathrm{H}, 4.3 ; \mathrm{N}, 4.4 . \mathrm{C}_{34} \mathrm{H}_{25} \mathrm{ClN}_{2} \mathrm{O}_{7}$ requires C, $67.05 ; \mathrm{H}, 4.14 ; \mathrm{N}$, $4.60 \%) ; \delta_{\mathrm{C}}\left(\mathrm{CDCl}_{3}\right) 63.83\left(\mathrm{C}-5^{\prime}\right), 72.66,75.59,80.73,82.55\left(\mathrm{C}-1^{\prime}\right.$, $-2^{\prime},-3^{\prime}$, and $-4^{\prime}$ ), $128.24-133.50$ (C-5, $-7,-8$, and -Ar), 136.08 (C-

Table 1. ${ }^{1} \mathrm{H}$ N.m.r. chemical shifts ( $\delta$ ) of certain $C$-nucleosides

| Compound | Solvent ${ }^{\text {a }}$ | $1^{\prime}-\mathrm{H}$ | 2'-H | 3'-H |  | $4^{\prime}-\mathrm{H}$ | 5'-Ha | $5^{\prime}-\mathrm{Hb}$ | 3-H |  | Other |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| (3a) | A | 5.62(d) | 6.20(t) | 5.99(t) |  | 4.88(m) | 4.63(dd) | 4.88 (m) | 9.06(s) |  | 6-8.10(m, $5-\mathrm{H}, 7-\mathrm{H}, 8-\mathrm{H}$, Ar-H) |
| (3b) | A | 5.62(d) | 6.19(t) | 6.02(t) |  | 4.87(m) | 4.62 (dd) | 4.87 (m) | 9.04(s) |  | - 8.11(m, 5-H, 6-H, 8-H, Ar-H) |
| (3c) | A | 5.66(d) | 5.97(t) | 6.22(t) |  | 4.93(m) | 4.64(dd) | 4.93(m) | 9.23(s) |  | $\begin{aligned} & 6-8.02(\mathrm{~m}, \mathrm{Ar}-\mathrm{H}), 8.09(\mathrm{~d}, 8-\mathrm{H}) \\ & 8(\mathrm{dd}, 7-\mathrm{H}), 8.94(\mathrm{~d}, 5-\mathrm{H}) \end{aligned}$ |
| (3d) | B | 5.35(d) | 4.31(t) | 4.19(t) |  | 4.13(q) | 3.77 (dd) | 3.91 (dd) | 9.14(s) |  | $\begin{aligned} & 4(\mathrm{dd}, 7-\mathrm{H}), 8.06(\mathrm{~d}, 8-\mathrm{H}) \\ & 9(\mathrm{~d}, 5-\mathrm{H}) \end{aligned}$ |
| (3e) | C | 5.09(d) | 4.31(t) | 4.19(t) |  | 4.14(q) | 3.77(dd) | 3.91 (dd) | 9.23(s) |  | $\begin{aligned} & 9(\mathrm{dd}, 6-\mathrm{H}), 8.14(\mathrm{~d}, 5-\mathrm{H}), \\ & 5(\text { apparent s, } 8-\mathrm{H}) \end{aligned}$ |
| (df) | B | 5.13(d) | 4.34(t) |  | 4.15(m) |  | 3.78(dd) | 3.91 (dd) | 9.33(s) |  | $\begin{aligned} & 8(\mathrm{~d}, 8-\mathrm{H}), 8.59(\mathrm{dd}, 7-\mathrm{H}), \\ & 7(\mathrm{~d}, 5-\mathrm{H}) \end{aligned}$ |
| (5a) | A | 5.86(d) | 6.02 (dd) | ) $5.81(\mathrm{dd})$ |  | 4.86(m) | 4.71(m) | 4.86(m) | 9.04(s) |  | $\begin{aligned} & 6-8.10(\mathrm{~m}, \text { Ar-H), } 7.68(\mathrm{dd}, 7-\mathrm{H}), \\ & 5(\mathrm{~d}, 5-\mathrm{H}), 8.47(\mathrm{~d}, 8-\mathrm{H}) \end{aligned}$ |
| (5b) | A | 5.87(d) | 6.01 (dd) | d) $5.82(\mathrm{t})$ |  | 4.87(m) | 4.71(dd) | 4.87(m) | 9.01(s) |  | $\begin{aligned} & 0-7.61,7.90-8.11(\mathrm{~m}, 5-\mathrm{H}, \mathrm{Ar}-\mathrm{H}), \\ & 7(\mathrm{dd}, 6-\mathrm{H}), 8.53(\mathrm{~d}, 8-\mathrm{H}) \end{aligned}$ |
| (5c) | E | 5.48(3) | 4.31(t) | 4.21(t) |  | 4.11(q) | 3.91(dd) | 4.05 (dd) | 9.27(s) |  | $\begin{aligned} & 4(\mathrm{dd}, 7-\mathrm{H}), 8.15(\mathrm{~d}, 5-\mathrm{H}), \\ & 3(\mathrm{~d}, 8-\mathrm{H}) \end{aligned}$ |
| (5d) | B | 4.98(d) | 4.27(t) |  | 4.13(m) |  | 3.75(dd) | 3.90 (dd) | 8.88(s) |  | $\begin{aligned} & 9(\mathrm{dd}, 6-\mathrm{H}), 8.11(\mathrm{~d}, 5-\mathrm{H}), \\ & 3(\mathrm{~d}, 8-\mathrm{H}) \end{aligned}$ |
| (6a) | A | 5.47(d) | 6.15(t) | 5.89(t) |  | 4.85(m) | 4.64(dd) | 4.85(m) | 8.61(s) |  | $\begin{aligned} & 9-7.68,7.95-8.09(\mathrm{~m}, 8-\mathrm{H}, \mathrm{Ar}-\mathrm{H}), \\ & 1(\mathrm{dd}, 7-\mathrm{H}), 8.51(\mathrm{~d}, 5-\mathrm{H}) \end{aligned}$ |
| (6b) | A | 5.45(d) | 6.12(t) | 5.91 (t) |  | 4.87(m) | 4.63(dd) | 4.87(m) | 8.57(s) |  | $\begin{aligned} & 3-8.07(\mathrm{~m}, \mathrm{Ar}-\mathrm{H}), 7.64(\mathrm{dd}, 6-\mathrm{H}), \\ & 2(\mathrm{~d}, 8-\mathrm{H}), 8.44(\mathrm{~d}, 5-\mathrm{H}) \end{aligned}$ |
| (6c) | C | 4.85(d) | 5.06(dd) | ) $5.30(\mathrm{t})$ |  | 4.12(m) | 3.96 |  | 8.90(s) |  | $\begin{aligned} & 6(\mathrm{dd}, 7-\mathrm{H}), 8.14(\mathrm{~d}, 8-\mathrm{H}), \\ & 3(\mathrm{~d}, 5-\mathrm{H}) \end{aligned}$ |
| (6d) | B | 4.88(d) | 4.18(t) |  | 3.87(m) |  | 3.67(dd) | 3.78(dd) | 8.77(s) |  | $\begin{aligned} & 0(\mathrm{dd}, 6-\mathrm{H}), 8.04(\mathrm{~d}, 8-\mathrm{H}), \\ & 0(\mathrm{~d}, 5-\mathrm{H}) \end{aligned}$ |
| Compound | Solvent ${ }^{\text {a }}$ | $1^{\prime}-\mathrm{H}$ | 2'-H | 3'-H | $4^{\prime}-\mathrm{H}$ | $5^{\prime}-\mathrm{Ha}$ | $5^{\prime}-\mathrm{Hb}$ | 2-H, 3 |  | H-4 | Other |
| (4a) | A | 5.88(d) | 6.35(dd) 5 | 5.99 (dd) 4. | 4.88(q) | ) 4.60 (dd) | ) 4.78 (dd) | 6.86(d), 7.0 | 2(d) | 9.04(s) | $\begin{aligned} & 7.29-8.02(\mathrm{~m}, 6-\mathrm{H}, 8-\mathrm{H}, \mathrm{Ar}- \\ & \mathrm{H}), \end{aligned}$ |
| (4b) | A | 5.92(d) | $6.38(\mathrm{dd}) 6$ | 6.00 (dd) 5.0 | 5.01 (q) | ) $4.65(\mathrm{dd})$ | 4.85(dd) | 6.97(d), 7 | 2(d) | $8.86(\mathrm{~s})$ | $\begin{aligned} & 8.42(\mathrm{~d}, 9-\mathrm{H}) \\ & 7.26-8.09(\mathrm{~m}, 6-\mathrm{H}, \mathrm{Ar}-\mathrm{H}), \\ & 8.31(\mathrm{dd}, 7-\mathrm{H}), 9.54(\mathrm{~d}, 9-\mathrm{H}) \end{aligned}$ |
| (4c) | C | 5.19(d) | 4.39(t) | 4.03(m) |  | $3.17-$ | -3.49(m) | 7.02(d), 7. | 3(d) | 8.90(s) | $\begin{aligned} & 7.66(\mathrm{dd}, 8-\mathrm{H}), 7.92(\mathrm{~d}, 6-\mathrm{H}), \\ & 8.46(\mathrm{~d}, 9-\mathrm{H}) \end{aligned}$ |
| (4d) | D | 5.30(d) | $\begin{aligned} & 4.35- \\ & 4.91(\mathrm{~m}) \end{aligned}$ | 4.30 (m) |  | 3.83(app | parent d) | 7.18(d), 7.2 | 4(d) | 8.89(s) | $\begin{aligned} & 8.05(\mathrm{~d}, 6-\mathrm{H}), 8.32(\mathrm{dd}, 7-\mathrm{H}), \\ & 9.60(\mathrm{~d}, 9-\mathrm{H}) \end{aligned}$ |

6), 140.17, and 142.69 (C-4a and -8a), 144.85 (C-3), 152.46 (C-2), and $165.39(\mathrm{C}=\mathrm{O})$.

Compound (3b) ( $385 \mathrm{mg}, 43 \%$ ); $R_{\mathrm{F}} 0.17$; syrup (Found: C, 66.8 ; $\mathrm{H}, 4.2 ; \mathrm{N}, 4.3 \%) ; \delta_{\mathrm{C}}\left(\mathrm{CDCl}_{3}\right) 63.71\left(\mathrm{C}-5^{\prime}\right), 72.66,75.59,80.79$, 82.43 (C-1', -2', $-3^{\prime}$, and $-4^{\prime}$ ), 128.41-133.44 (C-5, $-6,-8$, and -Ar), 136.08 (C-7), 140.87, and 141.87 (C-4a and -8a), 144.15 (C$3), 153.16(\mathrm{C}-2)$, and 165.33 , and $166.03(\mathrm{C}=\mathrm{O})$.

Compound (4a) ( $80 \mathrm{mg}, 9 \%$ ); $R_{\mathrm{F}} 0.15$; foam (Found: C, $67.8 ; \mathrm{H}$, 4.4; $\mathrm{N}, 4.5 . \mathrm{C}_{37} \mathrm{H}_{27} \mathrm{ClN}_{2} \mathrm{O}_{7} \cdot \frac{1}{2} \mathrm{H}_{2} \mathrm{O}$ requires $\mathrm{C}, 67.73 ; \mathrm{H}, 4.30, \mathrm{~N}$, $4.27 \%) ; \delta_{\mathrm{C}}\left(\mathrm{CDCl}_{3}\right) 63.89\left(\mathrm{C}-5^{\prime}\right), 72.72,73.60,75.23,80.62\left(\mathrm{C}-1^{\prime}\right.$, $-2^{\prime},-3^{\prime}$, and $-4^{\prime}$ ), 107.76 and 114.31 (C-2 and -3), 118.18 (C-9), $127.54-129.70$ (C-3a, -9a, and -Ar), 130.64 (C-7), 133.21 and 133.62 (C-6 and -8), 138.13 and 138.18 (C-1 and -5a), 146.78 (C4), and 165.21 and $166.03(\mathrm{C}=\mathrm{O})$.

6-Nitro-2-(2,3,5-tri-O-benzoyl- $\beta$-D-ribofuranosyl)quinoxaline (3c) and 8-Nitro-1-(2,3,5-tri-O-benzoyl- $\beta$-D-ribofuranosyl)pyrrolo $[1,2-\mathrm{a}]$ quinoxaline ( $\mathbf{4} \mathbf{b}$ ).-A solution of the pyranulose (1) ( $494 \mathrm{mg}, 0.89 \mathrm{mmol}$ ) and 1,2-diamino-4-nitrobenzene (2b) (163 $\mathrm{mg}, 1.06 \mathrm{mmol})$ in toluene ( 10 ml ) was heated at $90^{\circ} \mathrm{C}$ for 4 h , and then the solvent was evaporated off under reduced pressure. T.l.c. (chloroform-methanol, 99:1) showed that the light yellow syrup contained two major components ( $R_{\mathrm{F}} 0.37$ and 0.45 ). The mixture was separated by p.l.c. with hexane-ethyl acetate (3:1) as developer $(\times 3)$.

Compound (3c) $(89 \mathrm{mg}, 16 \%) ; R_{\mathrm{F}} 0.45$; yellow foam (Found: C,
63.4; $\mathrm{H}, 4.2 ; \mathrm{N}, 6.2 . \mathrm{C}_{34} \mathrm{H}_{25} \mathrm{~N}_{3} \mathrm{O}_{9} \cdot \frac{3}{2} \mathrm{H}_{2} \mathrm{O}$ requires $\mathrm{C}, 63.15 ; \mathrm{H}$, $4.36 ; \mathrm{N}, 6.45 \%) \delta_{\mathrm{C}}\left(\mathrm{CDCl}_{3}\right) 63.63\left(\mathrm{C}-5^{\prime}\right), 72.58\left(\mathrm{C}-2^{\prime}\right), 75.50(\mathrm{C}-$ $\left.3^{\prime}\right), 81.00\left(\mathrm{C}-4^{\prime}\right), 82.49\left(\mathrm{C}-1^{\prime}\right), 123.65$ (C-7), 125.74 (C-5), 128.39-134.60 (C-Ar), 131.20 (C-8), 141.25 (C-4a), 143.95 (C8a), 146.21 (C-3), 148.01 (C-6), 155.63 (C-2), and 165.35, 165.42, and $166.03(\mathrm{C}=\mathrm{O})$.

Compound (4b) ( $140 \mathrm{mg}, 43 \%$ ); $R_{\mathrm{F}} 0.37$; yellow foam (Found: C, 63.4; $\mathrm{H}, 4.2 ; \mathrm{N}, 6.2 . \mathrm{C}_{37} \mathrm{H}_{27} \mathrm{~N}_{3} \mathrm{O}_{9} \cdot \frac{1}{2} \mathrm{H}_{2} \mathrm{O}$ requires C, $63.15 ; \mathrm{H}$, 4.36; $\mathrm{N}, 6.50 \%$ ); $\delta_{\mathrm{C}}\left(\mathrm{CDCl}_{3}\right) 63.94\left(\mathrm{C}-5^{\prime}\right), 72.31,73.13,74.83$, and 81.08 ( $\mathrm{C}-1^{\prime},-2^{\prime},-3^{\prime}$, and $\left.-4^{\prime}\right), 109.28$ and $115.19(\mathrm{C}-2$ and -3$)$, 113.85 (C-9), 119.93 (C-7), 128.53-133.62 (C-1, -3a, -6, -9a, and -Ar), 141.02 and 141.46 (C-8 and -5a), 148.71 (C-4), and 165.21 , 166.03, and $166.50(\mathrm{C}=\mathrm{O})$.

6-Chloro-2-(2,3,5-tri-O-benzoyl- $\beta$-D-ribofuranosyl)quinoxaline 1-Oxide (5a) and 4-Oxide (6a).-To a solution of compound (3a) $(60 \mathrm{mg}, 0.1 \mathrm{mmol})$ in dichloromethane $(5 \mathrm{ml})$ at $0^{\circ} \mathrm{C}$ was added MCPBA ( $81.7 \mathrm{mg}, 0.4 \mathrm{mmol}$ ), and the mixture was kept at room temperature for 48 h . The solvent was evaporated off under reduced pressure. T.l.c. (benzene-methanol, 99:1) showed that the residue contained two major components ( $R_{\mathrm{F}}$ 0.36 and 0.35 ). The mixture was separated by p.l.c. with benzene-methanol (99:1) as developer ( $\times 5$ ).

Compound (5a) ( $12 \mathrm{mg}, 19 \%$ ); $R_{\mathrm{F}} 0.36$; foam (Found: C, 65.0 ; $\mathrm{H}, 4.1 ; \mathrm{N}, 4.3 . \mathrm{C}_{34} \mathrm{H}_{25} \mathrm{ClN}_{2} \mathrm{O}_{8}$ requires $\mathrm{C}, 65.33 ; \mathrm{H}, 4.03 ; \mathrm{N}$, $4.48 \%) ; \delta_{\mathrm{C}}\left(\mathrm{CDCl}_{3}\right) 63.63\left(\mathrm{C}-5^{\prime}\right), 71.90,73.90,78.69$, and 79.27




I
II
V
$\sqrt{1}$

III



IV
$\mid-\mathrm{H}_{2} \mathrm{O}$
(4b)
$R=$



VII
$\mid-\mathrm{MeCOCH}_{2} \mathrm{OH}$
(3c)

Scheme 2.
(C-1', -2', $-3^{\prime}$, and $-4^{\prime}$ ), 120.22 (C-8), 128.47-133.50 (C-2, -5, -7, and -Ar), 136.08 (C-6), 142.69 and 144.85 (C-4a and -8a), 145.91 ( $\mathrm{C}-3$ ), and $165.21(\mathrm{C}=\mathrm{O})$.

Compound ( $\mathbf{6 a}$ ) ( $37 \mathrm{mg}, 55 \%$ ); $R_{\mathrm{F}} 0.35$; foam (Found: C, 65.5 ; $\mathrm{H}, 4.1 ; \mathrm{N}, 4.55 \%) ; \delta_{\mathrm{C}}\left(\mathrm{CDCl}_{3}\right) 63.53(\mathrm{C}-5$ '), 72.43, 75.47, 80.79 , and 82.08 ( $\mathrm{C}-1^{\prime},-2^{\prime},-3^{\prime}$, and $-4^{\prime}$ ), 120.28 (C-5), 128.18-133.50 (C-3, -7, -8, and -Ar), 135.37 (C-6), 138.13, and 145.32 (C-4a and $-8 \mathrm{a}), 156.44(\mathrm{C}-2)$, and 165.21, 165.39, and $166.15(\mathrm{C}=\mathrm{O})$.

7-Chloro-2-(2,3,5-tri-O-benzyol- $\beta$-D-ribofuranosyl)quinoxa-
line 1-Oxide (5b) and 4-Oxide (6b).-The same procedure was used as for the reaction of ( $\mathbf{3 a}$ ) with MCPBA, but with substrate (3b).

Compound (5b) ( $17 \%$ ); $R_{\mathrm{F}} 0.32$; foam (Found: C, 65.4 ; H, 4.4; $\mathrm{N}, 4.2 \%) ; \delta_{( }\left(\mathrm{CDCl}_{3}\right) 63.42\left(\mathrm{C}-5^{\prime}\right), 71.96,73.95,78.63$, and 79.39 (C-1', $-2^{\prime},-3^{\prime}$, and -4'), 118.23 (C-8), 128.47-133.44 (C-2, -5, -6, and -Ar), 137.01 (C-7), 139.12, and 143.86 (C-4a and -8a), 144.79 ( $\mathrm{C}-3$ ), and 165.21, and $166.21(\mathrm{C}=\mathrm{O})$.

Compound (6b) ( $46 \%$ ); $R_{\mathrm{F}} 0.31$; foam (Found: C, 64.6; H, 4.3; $\mathrm{N}, 4.3 . \mathrm{C}_{34} \mathrm{H}_{25} \mathrm{ClN}_{2} \mathrm{O}_{8} \frac{1}{3} \mathrm{H}_{2} \mathrm{O}$ requires C, 64.71; $\mathrm{H}, 4.15 ; \mathrm{N}$,

Table 2. ${ }^{1} \mathrm{H}$ N.m.r. coupling constants $(\mathrm{Hz})$ of certain $C$-nucleosides

| Compound | $1^{\prime}, 2^{\prime}$ | $2^{\prime}, 3^{\prime}$ | $3^{\prime}, 4^{\prime}$ | $4^{\prime}, 5^{\prime} \mathrm{a}$ | $4^{\prime}, 5^{\prime} \mathrm{b}$ | $5^{\prime} \mathrm{a}, 5^{\prime} \mathrm{b}$ | 5,6 | 5,7 | 6,8 | 7,8 |  |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| (3a) | 4.7 | 4.7 | 4.7 | 3.7 | $a$ | 11.8 |  | $a$ |  | $a$ |  |
| (3b) | 5.0 | 5.0 | 5.0 | 3.7 | $a$ | 8.1 | $a$ |  | $a$ |  |  |
| (3c) | 4.7 | 4.7 | 4.7 | 3.7 | $a$ | 12.1 |  | 2.6 |  | 9.2 |  |
| (3d) | 5.3 | 5.3 | 5.3 | 3.7 | 3.0 | 12.1 |  | 2.4 |  | 9.1 |  |
| (3e) | 5.3 | 5.3 | 5.3 | 4.0 | 3.3 | 12.1 | 9.1 |  | 2.4 |  |  |
| (3f) | 5.7 | 5.7 | $a$ | 4.0 | 3.0 | 11.8 |  | 2.4 |  | 9.4 |  |
| (5a) | 3.7 | 5.7 | 7.4 | $a$ | $a$ | $a$ |  | 2.4 |  | 9.4 |  |
| (5b) | 3.7 | 3.7 | 3.7 | 4.0 | $a$ | 11.4 | 8.7 |  | 2.4 |  |  |
| (5c) | 3.4 | 3.4 | 3.4 | 3.4 | 3.4 | 8.1 |  | 2.0 |  | 9.4 |  |
| (5d) | 5.4 | 5.4 | 5.4 | 4.0 | 3.0 | 12.4 | 9.1 |  | 2.3 |  |  |
| (6a) | 4.7 | 4.7 | 4.7 | 3.4 | $a$ | 11.8 |  | 2.0 |  | 9.1 |  |
| (6b) | 4.7 | 4.7 | 4.7 | 3.7 | $a$ | 11.4 | 9.4 |  | 2.4 |  |  |
| (6c) | 4.7 | 4.0 | 4.7 | $a$ | $a$ | $a$ |  | 2.4 |  | 9.1 |  |
| (6d) | 5.0 | 5.0 | $a$ | 4.4 | 3.2 | 11.4 | 9.4 |  | 2.0 |  |  |
| Compound | $1^{\prime}, 2^{\prime}$ | $2^{\prime}, 3^{\prime}$ | $3^{\prime}, 4^{\prime}$ | $4^{\prime}, 5^{\prime} \mathrm{a}$ | $4^{\prime}, 5^{\prime} \mathrm{b}$ | $5^{\prime} \mathrm{a}, 5^{\prime} \mathrm{b}$ | 2, 3 | 6,7 | 6,8 | 7,9 | 8,9 |
| (4a) | 7.1 | 5.7 | 4.0 | 3.7 | 3.3 | 12.1 | 4.3 |  | $a$ |  | 9.0 |
| (4b) | 7.7 | 4.4 | 4.0 | 4.4 | 3.3 | 12.1 | 4.3 | 8.7 |  | 2.4 |  |
| (4c) | 7.0 | 4.3 | $a$ | $a$ | $a$ | $a$ | 4.0 |  | 2.4 |  | 9.1 |
| (4d) | 7.7 | $a$ | $a$ | $a$ | $a$ | $a$ | 4.0 | 9.0 |  | 2.3 |  |

${ }^{a}$ Unresolved.
$4.43 \%) ; \delta_{\mathrm{C}}\left(\mathrm{CDCl}_{3}\right) 63.59\left(\mathrm{C}-5^{\prime}\right), 72.37,75.47,80.67$, and $82.08(\mathrm{C}-$ $1^{\prime},-2^{\prime},-3^{\prime}$, and $\left.-4^{\prime}\right), 118.23$ (C-5), 128.41-133.50 (C-3, $-6,-8$, and -Ar), 136.60 (C-7), 138.30 , and 143.45 (C-4a and -8a), 155.32 (C$2)$, and $165.70,165.39$, and $166.15(\mathrm{C}=\mathrm{O})$.

General Deoxygenation Procedure.-A solution of a quinoxaline $N$-oxide and triphenylphosphine in methanol was heated under reflux for 3 h , and then the solvent was evaporated off under reduced pressure. The residue was purified by p.l.c with diisopropyl ether-methanol (93:7) as developer, to give the corresponding deoxygenated $C$-nucleoside. Identification was confirmed by comparison of i.r. and ${ }^{1} \mathrm{H}$ n.m.r. spectra with those of the deoxygenated $C$-nucleosides.

General Deprotection Procedure.-Sufficient methanolic sodium hydroxide was added to the protected $C$-nucleoside in absolute methanol. The mixture was kept at room temperature for 5 h , rendered neutral with acetic acid, and evaporated. The residue was purified by p.l.c. to afford the free $C$-nucleoside.
6 -Chloro-2-( $\beta$-D-ribofuranosyl)quinoxaline ( $\mathbf{3 d}$ ). This compound $(61 \%)$ was obtained from the tri- $O$-benzoate (3a), as crystals m.p. $178-181^{\circ} \mathrm{C}$ (from methanol) (Found C, $52.4 ; \mathrm{H}$, 4.1; $\mathrm{N}, 9.8 . \mathrm{C}_{13} \mathrm{H}_{13} \mathrm{ClN}_{2} \mathrm{O}_{4}$ requires C, $52.61 ; \mathrm{H}, 4.42 ; \mathrm{N}, 9.44 \%$ ); $\delta_{\mathrm{C}}\left(\mathrm{CD}_{3} \mathrm{OD}\right) 61.40\left(\mathrm{C}-5^{\prime}\right), 71.14,76.58,83.95$, and $85.30\left(\mathrm{C}-1^{\prime}\right.$, $-2^{\prime},-3^{\prime}$, and $-4^{\prime}$ ), 127.59, 130.58, and 130.81 (C-5, -7 , and -8 ), 134.03 (C-6), 139.30 and 141.69 (C-4a and -8a), 145.56 (C-3), and 156.38 (C-2).

7-Chloro-2-( $\beta$-D-ribofuranosyl)quinoxaline (3e). This compound ( $71 \%$ ) was obtained from the tri-O-benzoate ( $\mathbf{3 b}$ ), as crystals m.p. $145-147^{\circ} \mathrm{C}$ (from methanol) (Found: C, $52.8 ; \mathrm{H}$, 4.3; N, 9.6\%); $\delta_{\mathrm{C}}\left(\mathrm{CD}_{3} \mathrm{OD}\right) 61.31$ (C-5'), 71.14, 76.76, 84.01, and 85.24 (C-1', $-2^{\prime},-3^{\prime}$, and $-4^{\prime}$ ), 127.42, 130.34, and 130.70 (C-5, -6, and -8), 134.56 (C-7), 140.00 , and 140.99 (C-4a and -8a), 144.91 (C-3), and 156.96 (C-2).

6 -Nitro-2-( $\beta$-D-ribofuranosyl)quinoxaline (3f). This compound ( $25 \%$ ) was obtained from the tri- $O$-benzoate ( $\mathbf{3 c}$ ), as pale yellow crystals m.p. $253-255^{\circ} \mathrm{C}$ (from methanol) (Found: C, $50.9 ; \mathrm{H}, 4.35 ; \mathrm{N}, 14.0 . \mathrm{C}_{13} \mathrm{H}_{13} \mathrm{~N}_{3} \mathrm{O}_{6}$ requires C, $50.81 ; \mathrm{H}, 4.26 ; \mathrm{N}$, $13.68 \%$ ).

7-Chloro-1-( $\beta$-D-ribofuranosyl) pyrrolo[1,2-a]quinoxaline
( $\mathbf{4 c}$ ). This compound ( $97 \%$ ) was obtained from the tri-Obenzoate (4a), as needles m.p. $196-199^{\circ} \mathrm{C}$ (from methanol)
(Found: C, 51.9 ; $\mathrm{H}, 5.4 ; \mathrm{N}, 7.4 . \mathrm{C}_{16} \mathrm{H}_{15} \mathrm{ClN}_{2} \mathrm{O}_{4} \cdot 2 \mathrm{H}_{2} \mathrm{O}$ requires $\mathrm{C}, 51.83 ; \mathrm{H}, 5.17 ; \mathrm{N}, 7.56 \%$ ) $\delta_{\mathrm{C}}\left[\left(\mathrm{CD}_{3}\right)_{2} \mathrm{SO}\right] 61.68$ (C-5'), 71.05 (C-3'), 73.75 (C-2'), 75.55 (C-1'), 85.31 (C-4'), 107.89, and 114.53 (C-2 and -3), 119.07 (C-9), 127.32 (C-9a), 127.42 (C-3a), 127.64 (C-8), $128.30(\mathrm{C}-6), 129.06(\mathrm{C}-7), 131.26(\mathrm{C}-1), 137.93(\mathrm{C}-5 \mathrm{a})$, and 147.12 (C-4).

8-Nitro-1-( $\beta$-D-ribofuranosyl) pyrrolo[1,2-a]quinoxaline (4d). This compound ( $30 \%$ ) was obtained from the tri- $O$-benzoate (4b), as yellow needles m.p. 254-255 ${ }^{\circ} \mathrm{C}$ (from methanol) (Found: C, 53.0; $\mathrm{H}, 4.8 ; \mathrm{N}, 11.6 . \mathrm{C}_{16} \mathrm{H}_{15} \mathrm{~N}_{3} \mathrm{O}_{6} \cdot \mathrm{H}_{2} \mathrm{O}$ requires C , $52.89 ; \mathrm{H}, 4.72 ; \mathrm{N}, 11.57 \%)$; $\left.\delta_{\mathrm{C}}\left[\left(\mathrm{CD}_{3}\right)_{2} \mathrm{SO}\right)\right] 61.54\left(\mathrm{C}-5^{\prime}\right), 70.84$ (C-3'), 73.45 (C.2'), 74.94 (C-1'), 85.97 (C-4'), 109.38, and 115.16 (C-2 and -3), 113.95 (C-9), 119.71 (C-7), 127.55 (C-3a), 127.83 (C9a), 130.22 (C-6), 131.92 (C-1), 141.14 (C-5a), 145.18 (C-8), and 148.94 (C-4).

6-Chloro-2-( $\beta$-D-ribofuranosyl)quinoxaline 1-Oxide (5c). This compound ( $84 \%$ ) was obtained from the tri- $O$-benzoate (5a), as crystals m.p. $165-166^{\circ} \mathrm{C}$ (from methanol) \{Found: $[M+\mathrm{H}]^{+}$ (f.a.b.), 313.0542. $\mathrm{C}_{13} \mathrm{H}_{14} \mathrm{ClN}_{2} \mathrm{O}_{5}$ requires $\left.M+\mathrm{H}, 313.0591\right\}$.

6 -Chloro-2-( $\beta$-D-ribofuranosyl) quinoxaline 4 -Oxide ( 6 c ). This compound ( $88 \%$ ) was obtained from the tri- $O$-benzoate ( 6 a ), as crystals m.p. 204-206 ${ }^{\circ} \mathrm{C}$ (from methanol) \{Found: $[M+\mathrm{H}]^{+}$ (f.a.b.), 313.0584. $\mathrm{C}_{13} \mathrm{H}_{14} \mathrm{ClN}_{2} \mathrm{O}_{5}$ requires $\left.M+\mathrm{H}, 313.0591\right\}$.

7-Chloro-2-( $\beta$-D-ribofuranosyl)quinoxaline 1-Oxide (5d). This compound $(82 \%$ ) was obtained from the tri- $O$-benzoate (5b), as crystals m.p. $182-183^{\circ} \mathrm{C}$ (from methanol) \{Found: $[M+\mathrm{H}]^{+}$(f.a.b.), 313.0577. $\mathrm{C}_{13} \mathrm{H}_{14} \mathrm{ClN}_{2} \mathrm{O}_{5}$ requires $M+\mathrm{H}$, $313.0591\}$.

7-Chloro-2-( $\beta$-D-ribofuranosyl) quinoxaline 4-Oxide (6d). This compound $(87 \%$ ) was obtained from the tri- $O$-benzoate ( $\mathbf{6 b}$ ), as crystals m.p. $144-146^{\circ} \mathrm{C}$ (from benzene-hexane, 1:1) \{Found: $[M+\mathrm{H}]^{+}$(f.a.b.), 313.0583. $\mathrm{C}_{13} \mathrm{H}_{14} \mathrm{ClN}_{2} \mathrm{O}_{5}$ requires $M+H$, $313.0591\}$.

General Acetonization Procedure.-To a solution of a deprotected $C$-nucleoside in acetone was added acetone containing toluene- $p$-sulphonic acid monohydrate and the mixture was kept at room temperature for 2 h . The reaction mixture was neutralized with sodium hydrogen carbonate and stirred for 15 min . The solid was collected by filtration and thoroughly washed with acetone. The filtrate and washings were combined, and evaporated under reduced pressure to give a syrup, which
was purified by p.l.c. with chloroform-methanol (97:3) as developer.
6-Chloro-2-(2,3-O-isopropylidene- $\beta$-D-ribofuranosyl)-quinoxaline ( $\mathbf{3 g}$ ). This compound ( $49 \%$ ) was obtained from the deprotected nucleoside ( $\mathbf{3 d}$ ), as a foam; $R_{\mathrm{F}} 0.32$ (chloroformmethanol, 97:3); $\delta_{\mathrm{H}}\left(\mathrm{CDCl}_{3}\right) 1.39$ and $1.67(6 \mathrm{H}$, each s , isopropylidene Me$), 3.75\left(1 \mathrm{H}, \mathrm{m}, 5^{\prime}-\mathrm{H}_{\mathrm{a}}\right), 4.02(1 \mathrm{H}, \mathrm{dd}, J 2.3$ and $\left.12.1 \mathrm{~Hz}, 5^{\prime}-\mathrm{H}_{\mathrm{b}}\right), 4.22\left(1 \mathrm{H}, \mathrm{q}, 4^{\prime}-\mathrm{H}\right), 4.96\left(2 \mathrm{H}, \mathrm{m}, 2^{\prime}-\right.$ and $\left.3^{\prime}-\mathrm{H}\right)$, $5.35\left(1 \mathrm{H}, \mathrm{d}, J 3.0 \mathrm{~Hz}, 1^{\prime}-\mathrm{H}\right), 7.75(1 \mathrm{H}, \mathrm{dd}, J 2.3$ and $9.1 \mathrm{~Hz}, 7-\mathrm{H})$, $8.02(1 \mathrm{H}, \mathrm{d}, J 9.1 \mathrm{~Hz}, 8-\mathrm{H}), 8.14(1 \mathrm{H}, \mathrm{d}, J 2.3 \mathrm{~Hz}, 5-\mathrm{H})$, and 8.92 ( $1 \mathrm{H}, \mathrm{s}, 3-\mathrm{H}$ ).

7-Chloro-2-(2,3-O-isopropylidene- $\beta$-D-ribofuranosyl)-quinoxaline ( $\mathbf{3 h}$ ). This compound ( $52 \%$ ) was obtained from the deprotected nucleoside (3e), as a foam; $R_{\mathrm{F}} 0.32$ (chloroformmethanol, $97: 3) ; \delta_{\mathrm{H}}\left(\mathrm{CDCl}_{3}\right) 1.39$ and $1.67(6 \mathrm{H}$, each s, isopropylidene Me$), 3.72\left(1 \mathrm{H}, \mathrm{m}, 5^{\prime}-\mathrm{H}_{\mathrm{a}}\right), 4.02(1 \mathrm{H}, \mathrm{dd}, J 3.7$ and $12.4 \mathrm{~Hz}, 5^{\prime}-\mathrm{H}_{\mathrm{b}}$ ), $4.22\left(1 \mathrm{H}, \mathrm{q}, 4^{\prime}-\mathrm{H}\right), 4.96\left(2 \mathrm{H}, \mathrm{m}, 2^{\prime}-\right.$ and $\left.3^{\prime}-\mathrm{H}\right)$, $5.35\left(1 \mathrm{H}, \mathrm{d}, J 4.3 \mathrm{~Hz}, 1^{\prime}-\mathrm{H}\right), 7.74(1 \mathrm{H}, \mathrm{dd}, J 2.0$ and $9.1 \mathrm{~Hz}, 6-\mathrm{H})$, $8.06-8.09(2 \mathrm{H}, \mathrm{m}, 5-\mathrm{and} 8-\mathrm{H})$, and $8.90(1-\mathrm{H}, \mathrm{s}, 3-\mathrm{H})$.

6-Nitro-2-(2,3-O-isopropylidene- $\beta$-D-ribofuranosyl) quinoxaline ( 3 i ). This compound ( $56 \%$ ) was obtained from the deprotected nucleoside ( $\mathbf{3 f}$ ), as a foam; $R_{\mathrm{F}} 0.30$ (chloroformmethanol, 99:1); $\delta_{\mathrm{H}}\left(\mathrm{CDCl}_{3}\right) 1.92$ and $2.12(6 \mathrm{H}$, each s , isopropylidene Me), $4.03-4.45\left(3 \mathrm{H}, \mathrm{m}, 4^{\prime}-\mathrm{H}\right.$ and $\left.5^{\prime}-\mathrm{H}_{2}\right), 4.62-$ $4.92\left(2 \mathrm{H}, \mathrm{m}, 2^{\prime}\right.$-and $\left.3^{\prime}-\mathrm{H}\right), 5.20\left(1 \mathrm{H}, \mathrm{d}, J 4.7 \mathrm{~Hz}, 1^{\prime}-\mathrm{H}\right), 8.64(1-\mathrm{H}$, d, $J 9.6 \mathrm{~Hz}, 8-\mathrm{H}), 8.96(1 \mathrm{H}$, dd, $J 2.3$ and $9.6 \mathrm{~Hz}, 7-\mathrm{H}), 9.33(1 \mathrm{H}$, d, $J 2.3 \mathrm{~Hz}, 5-\mathrm{H})$, and $9.66(1 \mathrm{H}, \mathrm{s}, 3-\mathrm{H})$.

7-Chloro-1-(2,3-O-isopropylidene- $\beta$-D-ribofuranosyl)pyrrolo-[1,2-a]quinoxaline (4e). This compound ( $43 \%$ ) was obtained
from the deprotected nucleoside ( $\mathbf{4 c}$ ), as a foam; $R_{\mathrm{F}} 0.33$ (chloroform); $\delta_{\mathrm{H}}\left(\mathrm{CDCl}_{3}\right) 1.37$ and $1.55(6 \mathrm{H}$, each s , isopropylidene Me), $3.18-3.50\left(2 \mathrm{H}, \mathrm{m}, 5^{\prime}-\mathrm{H}_{2}\right), 4.14\left(2 \mathrm{H}, \mathrm{m}, 3^{\prime}-\right.$ and $\left.4^{\prime}-\mathrm{H}\right), 4.90\left(1 \mathrm{H}, \mathrm{m}, 2^{\prime}-\mathrm{H}\right), 5.37\left(1 \mathrm{H}, \mathrm{d}, J 6.0 \mathrm{~Hz}, 1^{\prime}-\mathrm{H}\right), 7.12$ and $7.24(2 \mathrm{H}$, each d, $J 4.0 \mathrm{~Hz}, 2$ - and $3-\mathrm{H}), 7.47(1 \mathrm{H}, \mathrm{dd}, J 2.0$ and $8.1 \mathrm{~Hz}, 8-\mathrm{H}), 7.93(1 \mathrm{H}, \mathrm{d}, J 2.0 \mathrm{~Hz}, 6-\mathrm{H}), 8.42(1 \mathrm{H}, \mathrm{d}, J 8.1$ $\mathrm{Hz}, 9-\mathrm{H})$, and $8.92(1 \mathrm{H}, \mathrm{s}, 4-\mathrm{H})$.

8-Nitro-1-(2,3-O-isopropylidene- $\beta$-D-ribofuranosyl)pyrrolo-[1,2-a]quinoxaline ( $\mathbf{4 f}$ ). This compound ( $47 \%$ ) was obtained from the deprotected nucleoside ( $\mathbf{4 d}$ ), as a foam; $R_{\mathbf{F}} 0.33$ (chloroform); $\delta_{\mathrm{H}}\left(\mathrm{CDCl}_{3}\right) 1.39$ and $1.55(6 \mathrm{H}$, each s , isopropylidene Me), $3.18-3.51\left(4 \mathrm{H}, \mathrm{m}, 3^{\prime}-\right.$ and $4^{\prime}-\mathrm{H}$, and $\left.5^{\prime}-\mathrm{H}_{2}\right)$, $4.19\left(1 \mathrm{H}, \mathrm{m}, 2^{\prime}-\mathrm{H}\right), 5.35\left(1 \mathrm{H}, \mathrm{d}, J 6.9 \mathrm{~Hz}, 1^{\prime}-\mathrm{H}\right), 7.22$ and $7.35(2$ H, each d, $J 4.0 \mathrm{~Hz}, 2$-and $3-\mathrm{H}), 8.09(1 \mathrm{H}, \mathrm{d}, J 8.7 \mathrm{~Hz}, 6-\mathrm{H}), 8.34$ $(1 \mathrm{H}, \mathrm{dd}, J 2.7$ and $8.7 \mathrm{~Hz}, 7-\mathrm{H}), 9.07(1 \mathrm{H}, \mathrm{s}, 4-\mathrm{H})$, and $9.40(1 \mathrm{H}$, d, $J 2.7 \mathrm{~Hz}, 9-\mathrm{H}$ ).

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